

## How to Write a Chemistry II Lab Report

A lab report is one of a scientist's most valuable tools. It is a permanent record of everything that takes place in the laboratory. A properly written lab report contains unambiguous statements of what was done and observed by the scientist. It is a **factual** account of a researcher's work.

Our lab reports will be written in a bound notebook with numbered pages. Only the pre-lab (final draft), the data, the calculations, and the conclusion should be included in the lab book; **rough drafts should not be included in the bound notebook.**

Since the report is to be a permanent record, it is to be written **IN INK (blue or black only)** so it cannot be easily altered. Labs may not be re-written or typed, so BE NEAT the first time!

A chemistry lab report is an INDIVIDUAL report. You are to write it on your own, IN YOUR OWN WORDS. **For every section that is the same as your lab partner's (or anyone else's), you will both receive NO CREDIT (a score of zero).** NO COPYING!!

**DO NOT ERASE OR COVER MISTAKES!!!** Just draw a thin line through the mistake, make the change, and initial it. Do not erase mistakes, use "white out," or scribble over mistakes. "Mistakes" may be correct or may be of help later! You will not be marked down for correctly corrected mistakes! ☺

Miss Gergunis mistake	<del>Miss Gergunis</del> not correct	Miss Gergunis <sup>KG</sup> correct
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### Cover of Lab Book

On the cover of your lab book, **your name**, course (**Chemistry II**), and **hour** should be clearly marked. If you would like to decorate the outside of your lab book, that is fine as long as the above information is still clearly visible.

John Scientist Chemistry II 3rd hour
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### Numbered Pages

The pages in your lab book should be numbered **IN BLUE OR BLACK INK** to show that you did not add or remove pages during the course of a lab. This emphasizes that your work is your own and is what you actually did/observed during lab. Odd numbers are on the right-hand pages; left-hand pages should have even numbers. Every page in your lab book should be numbered.

RIGHT
1

LEFT	RIGHT
2	3

## Table of Contents

As the year progresses, we will do a lot of labs. It is nice to be able to quickly look at a table of contents to find where in your lab book you need to look. Your table of contents should be on page 1 of your lab book and should list the lab titles and the pages they begin on.

Example:

Table of Contents	
<u>Lab Title</u>	<u>Starts on Page</u>
Percent of Water in Alum Lab	3
Compound Formula Lab	10
Ratio of Moles in a Reaction Lab	18

The sections of a chemistry lab report are:

Title Page/Abstract -- started BEFORE LAB, completed AFTER LAB

Goal

Research

Materials

Hazards

Procedure

} "PRE-LAB" (completed BEFORE LAB)

Data -- written DURING LAB

Calculations (not always needed--skip it if not needed) -- written AFTER LAB

Conclusion -- written AFTER LAB

Each section should include a heading (such as **Hazards**), and the information under that heading should be BLOCK INDENTED (every line in that section indented, not just the first one).

### **Goal:**

The goal of this lab is to determine the boiling point of water in degrees Celsius.

### **Research:**

A boiling point is the temperature at which a pure substance changes from the liquid state to the gas state when heated. There are three common temperature scales: Celsius, Fahrenheit, and Kelvin. Fahrenheit is commonly used in the United States, but most of the rest of the world records temperatures in Celsius . . .

### **Materials:**

1 Celsius thermometer

1 25 x 150 mm test tube

1 Bunsen burner

1 sparker

1 heat resistant pad

1 ringstand

## Title Page

Each lab will have a title page. Title pages are always on odd (right-hand) pages. The title page should contain the lab title (the official title or one of your choice related to the lab), your name, partner's name, the course name, your hour, the date the lab was performed, and the lab abstract (see below). [For multiple-day labs, list all days.] The back of the title page is always left blank just in case your abstract runs long (but it shouldn't!).

Lab Title
Your Name Partner's Name Chemistry II nth hour
9/18/09
<b>Abstract:</b> The goal of this lab was to . . . (goal). During this lab, . . . ( <b>summary</b> of procedure). The third sentence is a <b>summary</b> of your data and calculations. My answer is . . . (give your answer).

## Goal

Each handout will list the goal of the lab. You may copy it word-for-word into your lab book or summarize it. Make sure your focus is always on meeting the lab's goal!

## Research

Before starting a lab, it is important to understand some of the ideas and chemistry behind what will happen. Each lab handout will include a list of topics to be researched before lab begins. (Most labs will also have problems to be tried as well.) These research topics are designed to help you understand what is going on in the lab.

Please note that your research section should be written in sentences and paragraphs (NOT A BULLETED LIST!!!). Equations should be written on their own lines and centered. Diagrams and calculations should also be included. **Put definitions and research IN YOUR OWN WORDS!** If there is information you must use word-for-word, it should be placed in quotations. List sources (where you find your information [title of book and page number, full website address, or person you spoke to]) at the end of each topic in parentheses. The more details and information included in the research, the better your grade will be! [NOTE: No research = no lab!]

**NOTE:** Current and former LHS students are not acceptable resources.

## Materials

The materials section of the lab report will list the materials, chemicals, and equipment you plan to use to complete the lab. It is written in **TWO (NOT ONE, NOT THREE)** columns that should NOT cross each other and should be approximately the same length. Materials are listed in the following format: **number required description/size/amount name of item/chemical**

Example:

Materials:

1 Bunsen burner	1 heat resistant pad
3 50 mL graduated cylinders	6.59 g $K_2Cr_2O_7$
25 mL 6 M HCl	2 #6 2-hole rubber stoppers
	4 25 x 150 mm test tubes

There are a few things that are NEVER included in a materials list: liquid water of any kind, paper towels, goggles, aprons, lab books, pens, lab desks, and lab partners.

### Hazards

Always begin your hazards section with the sentence “**The student safety contract applies.**” Then use complete sentences to mention the hazards of each chemical being used during the lab. Be sure to mention any special precautions or safety equipment needed (such as gloves or the explosion barrier). The safety contract is mentioned so you do not have to repeat the thirty safety rules every lab. Copies of MSDS sheets are available in the MSDS binders in the classroom or on the website [http://www.flinnsci.com/search\\_MSDS.asp](http://www.flinnsci.com/search_MSDS.asp)

### Procedure

Your procedure is your step-by-step list of how you plan to complete the lab. It should include cleaning equipment, setting up equipment, taking measurements, performing reactions, and even cleaning up. **It should NOT include doing calculations or writing the conclusion.** All procedures should include any diagrams you will need (especially of your set-up). **You CANNOT have your lab handout or rough draft at your desk during lab** (as it poses a fire hazard). Main idea steps get numbered; detail steps get letters. **If you have an “a,” you must have a “b.”** Make sure to clearly **label all set-up diagrams**. Each step should begin with a verb.

Procedure:

1. Clean equipment
  - a. physically clean glassware using test tube brush and tap water
  - b. chemically clean glassware using distilled water
  - c. dry equipment
  
2. Obtain 50 mL of HCl
  - a. label beaker HCl
  - b. take beaker to hood
  - c. put on gloves
  - d. pour HCl into beaker
  - e. measure 50 mL HCl using graduated cylinder

**The final step of any procedure should be “Clean up.”**

Never use I, we, you, me, or other pronouns in a procedure! Focus on what should be done, not who is doing it!

If a step is about setting up equipment or building something, a detailed, labeled diagram of what is being set up or built should be included next to that step in the procedure.

DO NOT write “get materials” or “record data.” Be more specific! Better examples would be “get 40 mL of HCl in 100 mL beaker” or “record height of flame.”

### Data

Data is the most crucial part of a lab report. It requires you to record something about everything you do in lab. More data = better data! Record colors, smells, shapes, and measurements! Be sure to use words, numbers, and sketches during every lab. Charts are also helpful; they may be set-up in your lab book before lab starts.

At the end of each day of lab, sign and date your data, have your partner(s) sign your data, and have the teacher initial that you have cleaned up your lab station. Any unused space on the page should be crossed out to show you will not add data after lab.

### Calculations (Skip this section for labs with no calculations)

The hardest thing to do in a calculations section is to make sure the reader can follow your math. Be sure to show which formulas you are using and where your numbers come from.

Example:

Calculations:

$$q = m \cdot \Delta t \cdot C_p$$

$$q = ?$$

$$m = 25.06 \text{ g (mass of my metal)}$$

$$\Delta t = \text{final temp} - \text{initial temp} = 26.0 \text{ }^\circ\text{C} - 100.0 \text{ }^\circ\text{C} = -74.0 \text{ }^\circ\text{C}$$

$$C_p = q / (m \cdot \Delta t) =$$

### Conclusion

Your conclusion tells the reader what you learned from the lab. It starts by summarizing the goal of the lab. It then gives the answer you figured out. Next, it explains how you figured out your answer by discussing your logic/reasoning, important data, and calculations (all to support your answer). It then explains any problems that occurred during your lab (if there were any). Finally, it lists at least one SPECIFIC way the lab could be improved (but “more time” will not get you any credit).

### Abstract

The abstract is the final section of a lab report. It is included on the title page. It is a **short** summary of the entire lab and report. It should only be 4-6 sentences long. Someone should be able to read your abstract and understand why you did the lab (the goal), what you did (a sentence about your procedure), what happened (a sentence that includes your key data and calculations), and what you found (your answer to the goal) without reading your entire report. A model of an abstract is shown on the example title page above.

## Lab Grades

- 1st** You must have a complete pre-lab (title page, goal, research, materials, hazards, and procedure) when you walk in for lab. Only pre-labs **in the lab book in blue or black ink** will be considered for grading. If your pre-lab is incomplete (missing a section, poorly done, not in your lab book, or not in blue or black ink), you will not be allowed to do the lab. If your pre-lab is incomplete, you may complete your pre-lab during class in the classroom while wearing goggles and sit out the lab.
- 2nd** Those who have completed pre-labs are allowed to do the lab. Those who do not will **not ever do the lab**. **There are no make-up labs if a person is unprepared** (has not completed his/her pre-lab, does not have his/her lab book, has written his/her pre-lab in pencil/red ink, does not have the proper attire/shoes, has not passed the safety quiz, has not brought in a signed safety contract, etc.)!!! Please note that if you arrive more than 10 minutes late for a lab, you may not do the lab.
- 3rd** Those who do the lab and turn in a completed lab report may received up to the maximum points. See the sample rubric below for the details:

<b>Percent of Water in a Hydrate Lab Scoring Rubric</b>					
Title Page, Abstract, and Goal:	2	1.6	1.4	1.2	0
All complete and correct	Incomplete/incorrect		Missing all three		
Research:	3	2.4	2.1	1.8	0
Research thorough, correct, detailed, in own words	Thorough research but with errors		Incomplete research	No research	
Materials:	2	1.6	1.4	1.2	0
Complete list	Incomplete or improper format			Missing	
Hazards:	1	0.8	0.7	0.6	0
Complete; in sentences	Incomplete; in fragments			Missing	
Procedure:	2	1.6	1.4	1.2	0
Complete and detailed	Incomplete or wrong format			Missing	
Data and Calculations:	5	4	3.5	3	0
Complete, detailed, correct	Incomplete/vague/incorrect		No calcs		No data
Conclusion:	3	2.4	2.1	1.8	0
Complete and correct answer	Incomplete/incorrect			Missing	
Housekeeping and Signatures:	2	1.6	1.4	1.2	0
Has all signatures	Missing some signatures		Missing all signatures		
TOTAL - Percent of Water in a Hydrate Lab				out of 20 points	